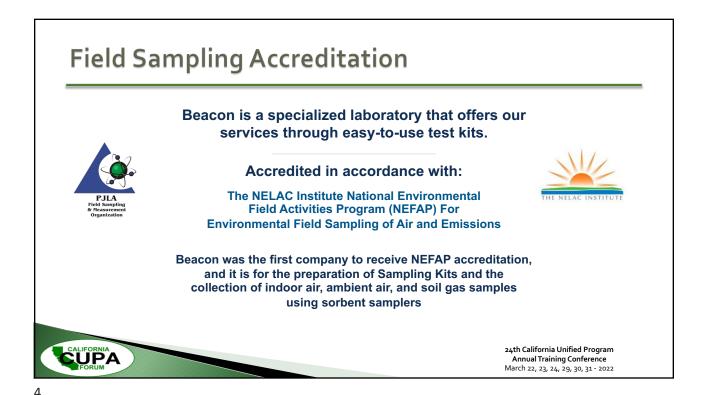
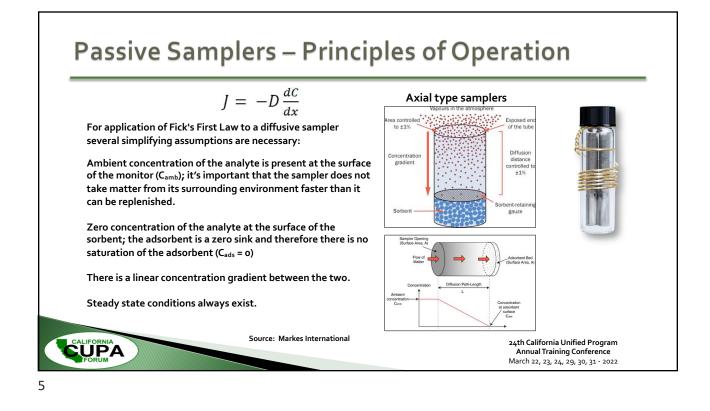
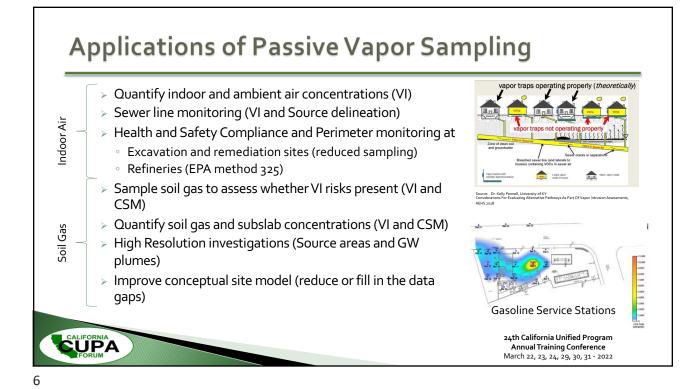


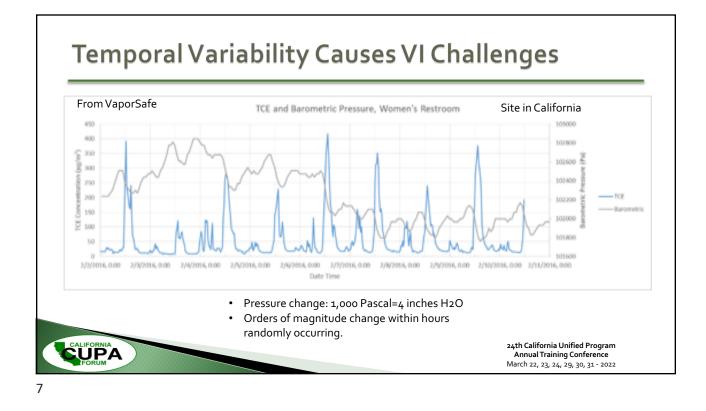
# <section-header> Outline Introductions Passive Sampling and Indoor Air Regulatory Guidance Open Intrusion Challenges we all deal with and why Passive is Preferred Open challenges for soil gas sampling Infuence of geological and evapotranspiration conditions Advection and diffusion in typical soil gas investigations Soil gas regulatory guidance and sampling methodology concerns Spatial variability during soil gas investigations Advancements in Passive Sampling

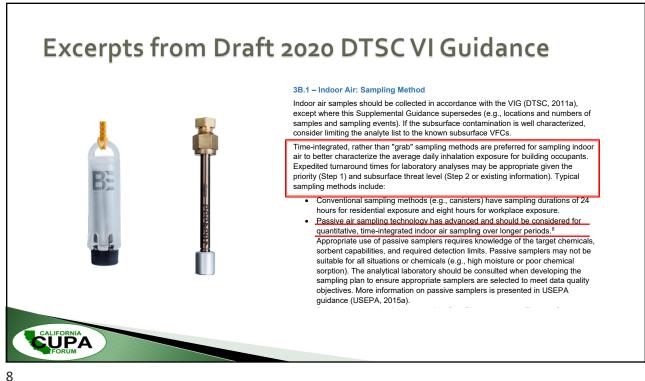


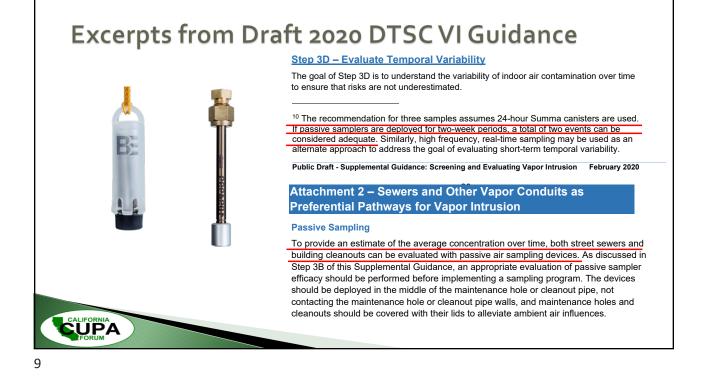


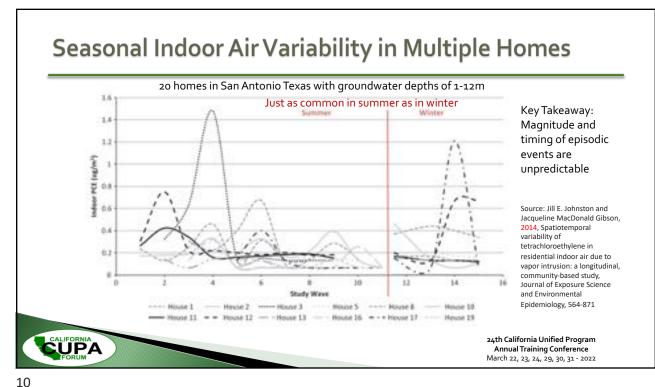




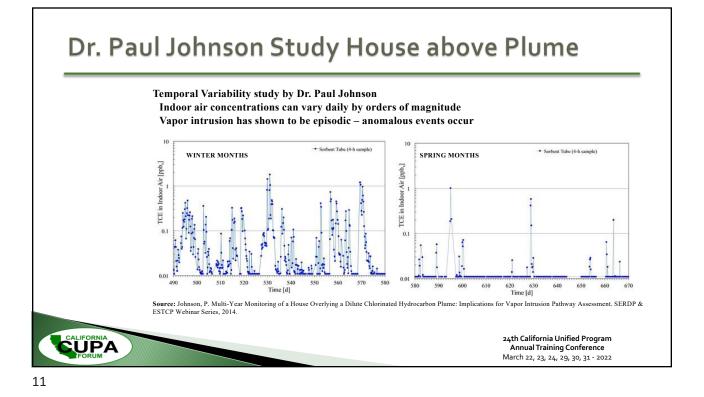


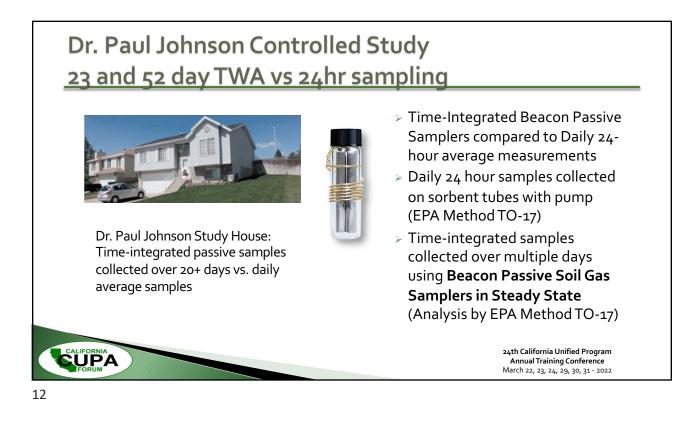




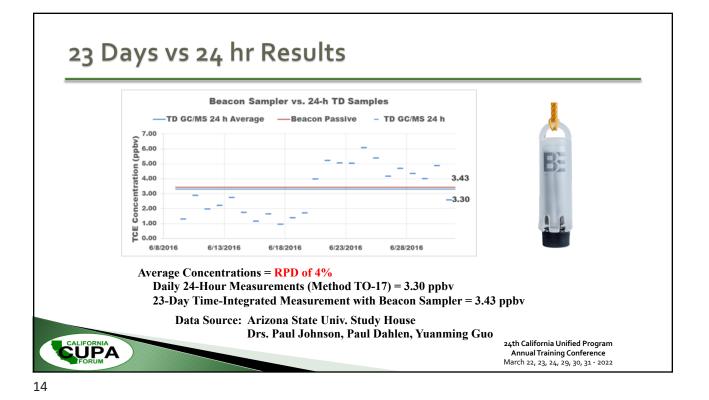


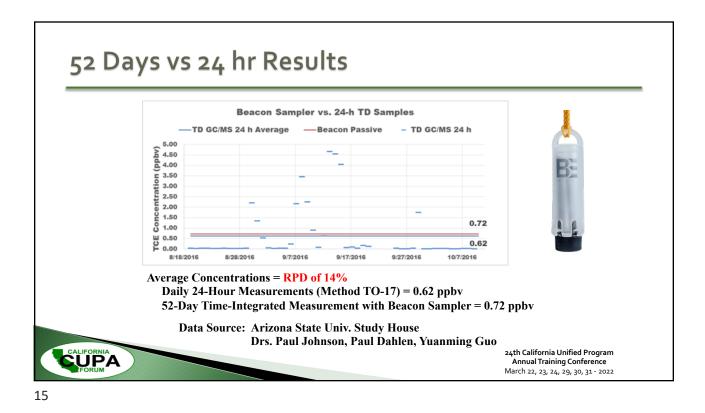
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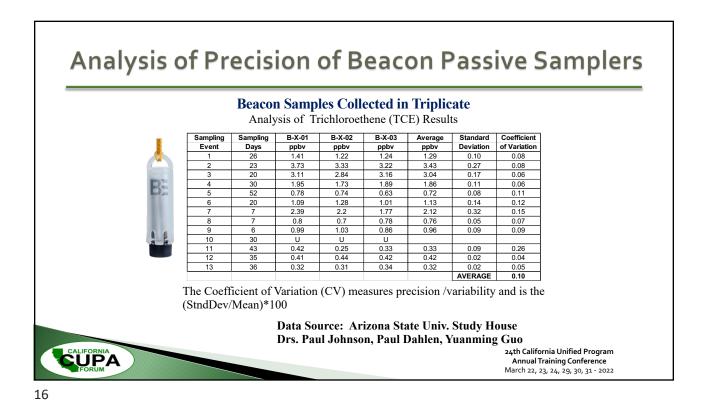


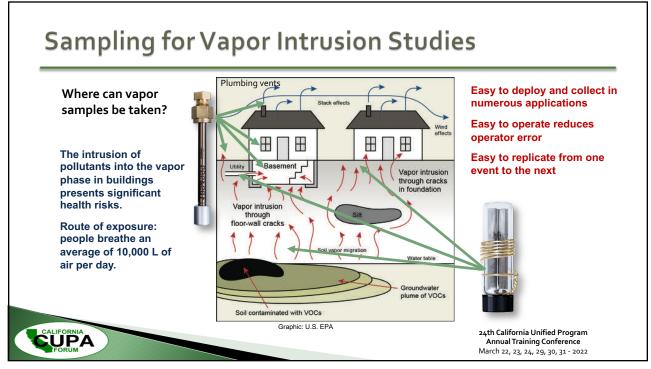


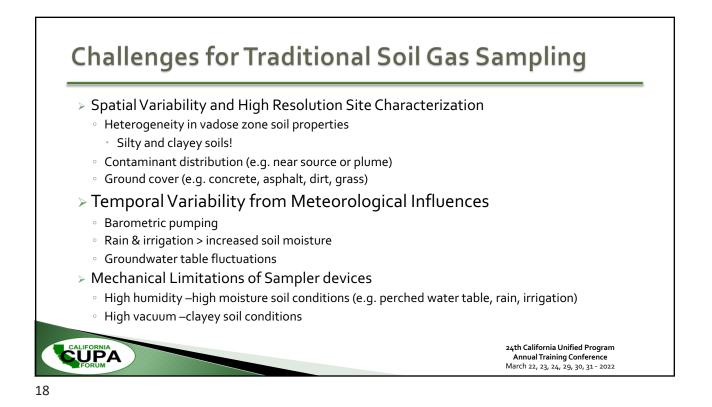
# Study Set-Up Beacon Passive Samplers were collected $\geq$ in triplicate; exposed for the duration of the sampling periods Pumped samples were collected at a flow ≻ rate of 10 ml/min with a total volume of 14.4 L per day Average concentrations were calculated from multiple 24hr day results 24th California Unified Program Annual Training Conference March 22, 23, 24, 29, 30, 31 - 2022

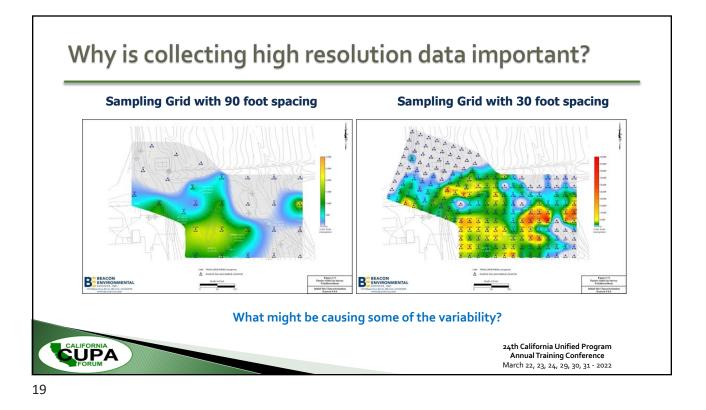


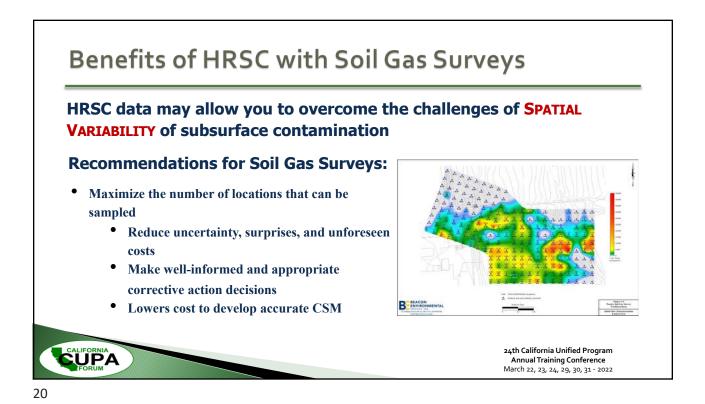


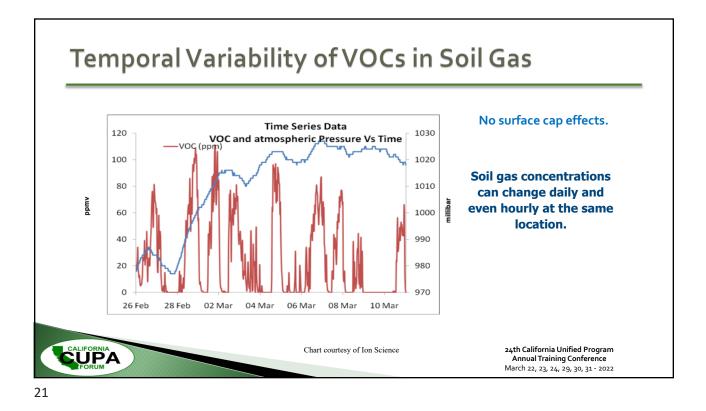


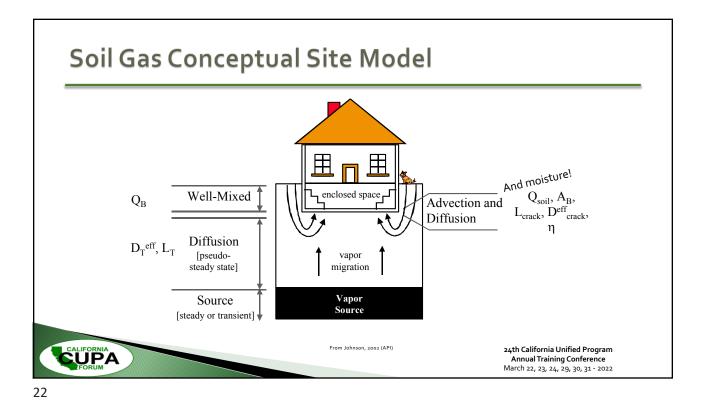


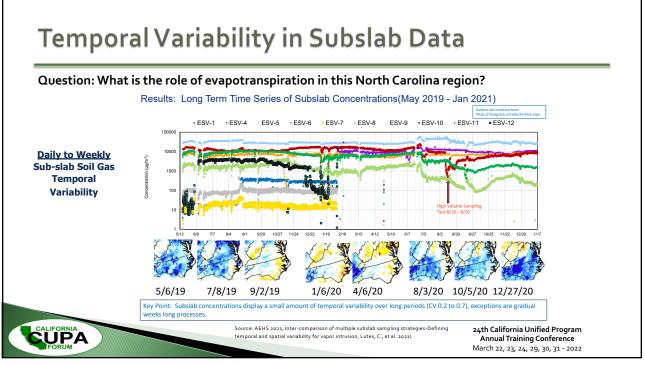


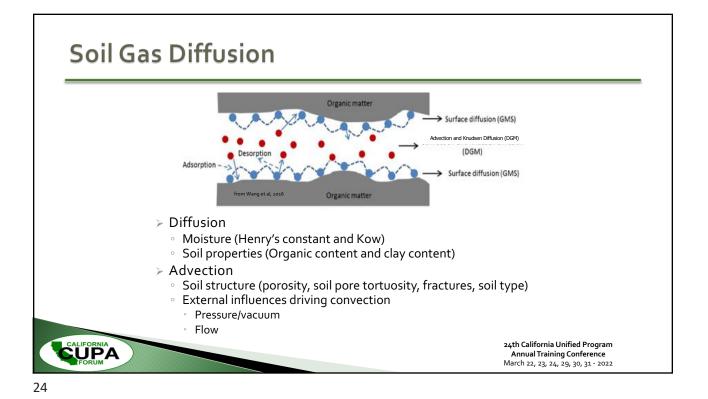


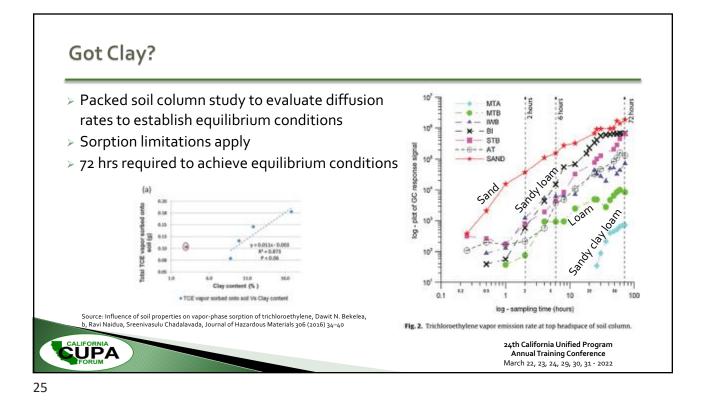












# Soil Gas Models and Moisture Conditions

### Millington-Quirk Model (1961)

- Calculates the effective diffusion coefficient across the capillary fringe where high moisture conditions prevail (EPA, J&E Model, V6, 2017).
- Defines conditions VOC diffusion under Steady-State inclusive of water moisture McAlary et al (2014).

### Effective molecular diffusion coefficient

inclusive of Millington-Quirk for gas and water phases

Utilized in Johnson and Ettinger and EPA VI models

$$D^{eff} = D^{air} \frac{\theta_v^{3.33}}{\theta_T^2} + \left(\frac{D^{H2O}}{H_i}\right) \frac{\theta_m^{3.33}}{\theta_T^2}$$

where

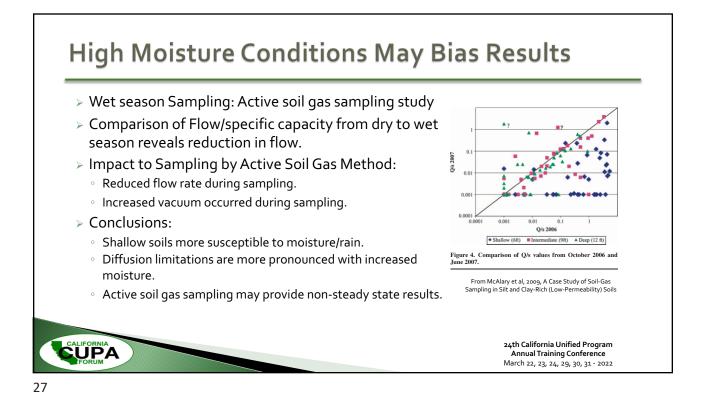
- ${\rm H_i} \quad = \quad the \ chemical-specific \ Henry's \ Law \ constant \ [(ug/m^3-vapor)/(ug/m^3-H_2O)]$
- $\theta_m$  = the volumetric moisture content [m<sup>3</sup>-H<sub>2</sub>O/m<sup>3</sup>-soil]
- $\theta_T$  = the total porosity [m<sup>3</sup>-voids/m<sup>3</sup>-soil]
- $\theta_V$  = the volumetric vapor content (= $\theta_T \theta_m$ ) [m<sup>3</sup>-vapor/m<sup>3</sup>-soil]
- $D^{air}$  = the chemical-specific molecular diffusion coefficient in air  $[m^2/d]$

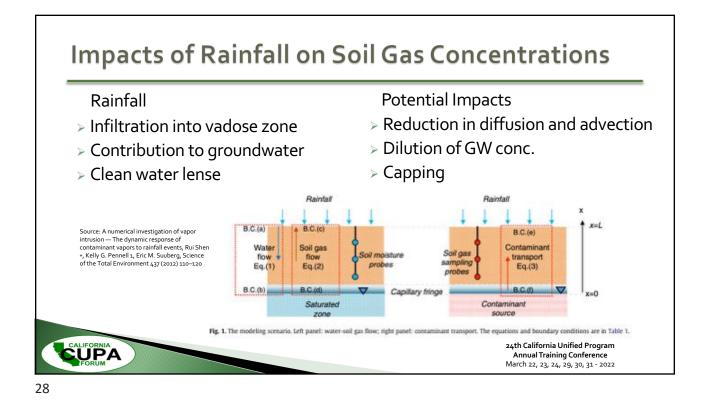
 $D^{\rm H2O} = ~~ \mbox{the chemical-specific molecular diffusion coefficient in water} \left[m^2/d\right]$ 

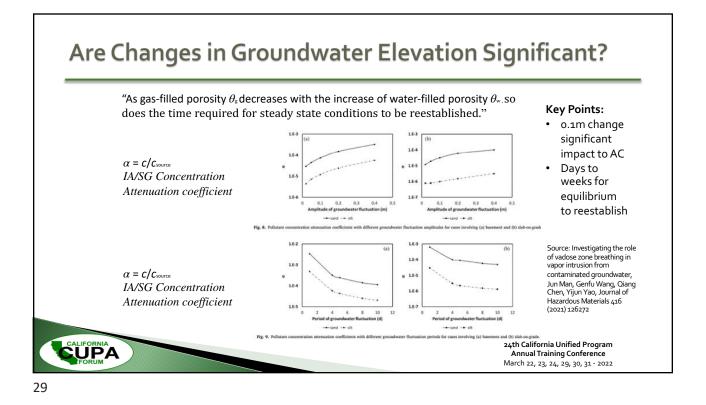
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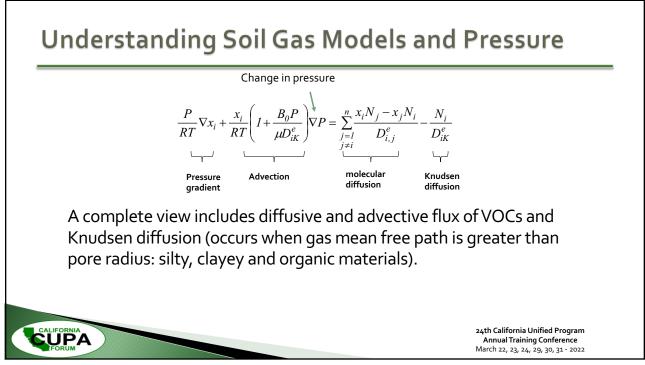


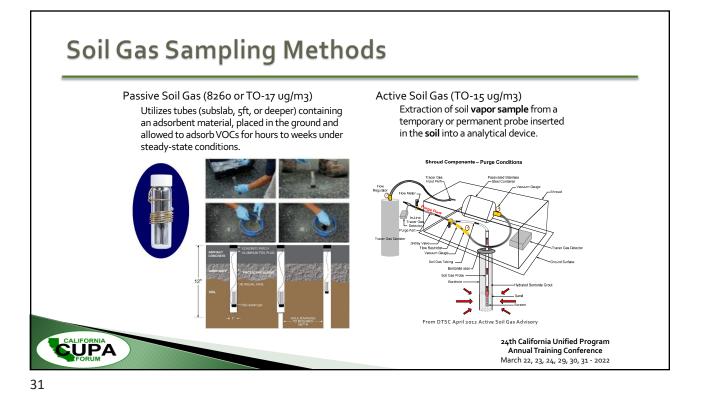
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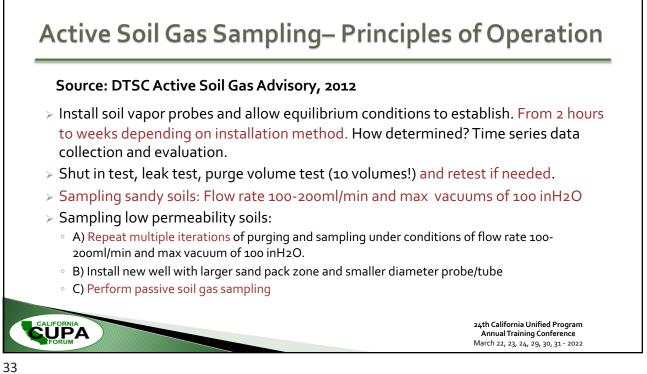




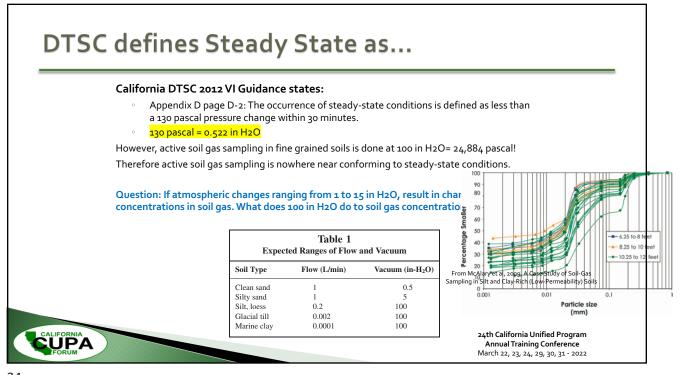


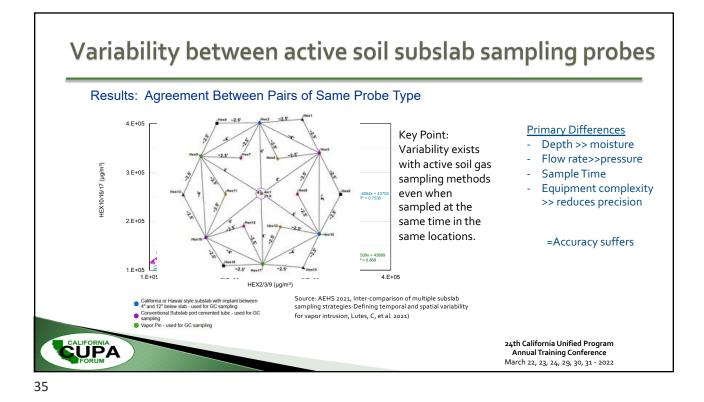


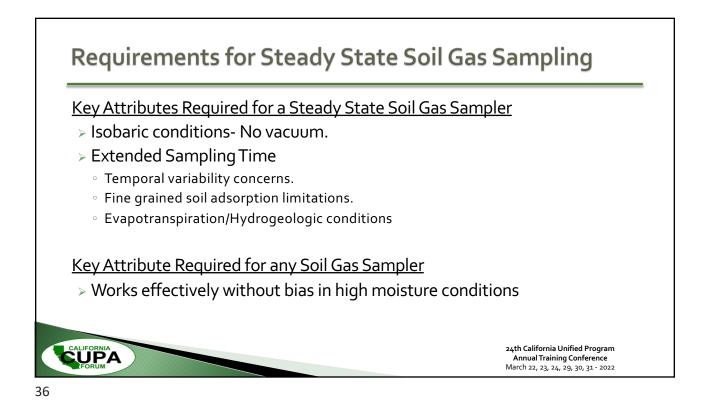
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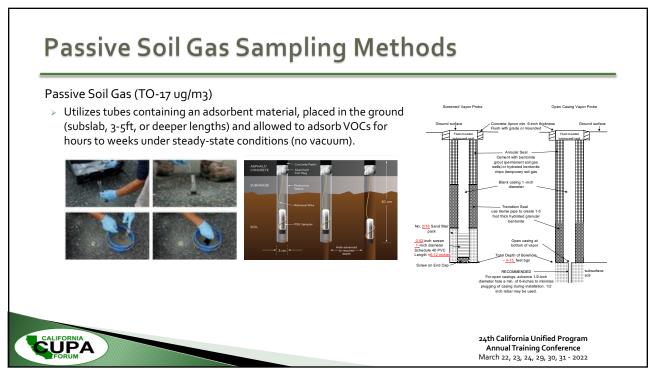






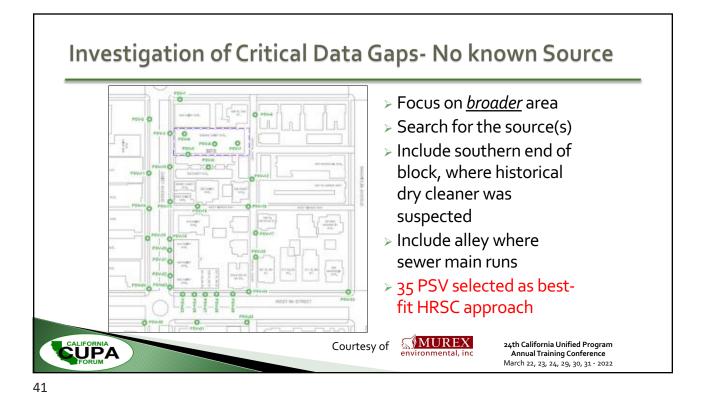


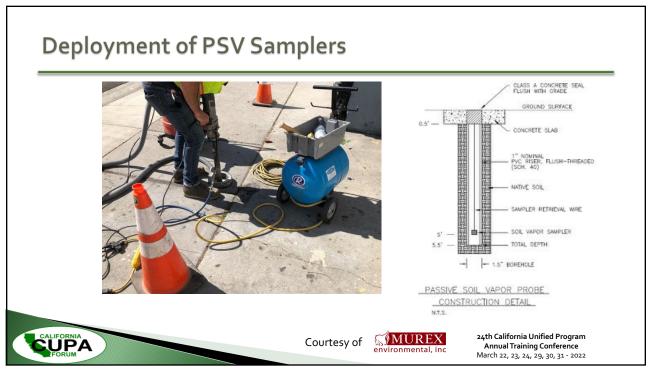
## **Beacon Passive Gas Samplers** Actual size: 18 mm x 60 mm (0.7 x 2.4 inches) • Hydrophobic Adsorbents Two types of adsorbents Two pairs of adsorbents for duplicates Uniform mass of adsorbents used (verified with analytical balance) **Completely inert sampler** • **Compliant with ASTM Standards** • D5314 and D7758 Report Concentration Data (ug/m<sup>3</sup>) 24th California Unified Program Annual Training Conference March 22, 23, 24, 29, 30, 31 - 2022 37



Longer Sampling Periods allow for:		Limits of Detection (ug/m <sup>3</sup> )				
	COMPOUND	1 Day	3 Days	7 Days	14 Days	26 Days
<ul> <li>Lower LODs</li> <li>Reduced influence of temporal variability</li> <li>Improved assessment of steady-state soil gas VOC concentrations</li> <li>Example Sampling Periods Possible:</li> <li>&lt;24 hours</li> <li>1-14 days</li> <li>Longer Sampling Periods possible.</li> </ul>	Vinyl Chloride	<4.29	<1.43	<0.61	<0.31	<0.16
	1,1-Dichloroethene	<10.52	<3.51	<1.50	<0.75	<0.40
	trans-1,2-Dichloroethe	ne <7.89	<2.63	<1.13	<0.56	<0.30
	1,1-Dichloroethane	<4.08	<1.36	<0.58	<0.29	<0.16
	cis-1,2-Dichloroethene	<6.55	<2.18	<0.94	<0.47	<0.25
	1,2-Dichloroethane	<6.20	<2.07	<0.89	<0.44	<0.24
	1,1,1-Trichloroethane	<3.31	<1.10	<0.47	<0.24	<0.13
	Trichloroethene	<10.52	<3.51	<1.50	<0.75	<0.40
	Tetrachloroethene	<8.47	<2.82	<1.21	<0.60	<0.33
	Benzene	<13.10	<4.37	<1.87	<0.94	<0.50
	Toluene	<17.36	<5.79	<2.48	<1.24	<0.67
	Ethylbenzene	<8.17	<2.72	<1.17	<0.58	<0.31
	Xylenes	<7.89	<2.63	<1.13	<0.56	<0.30







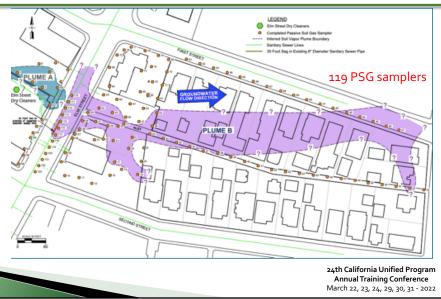


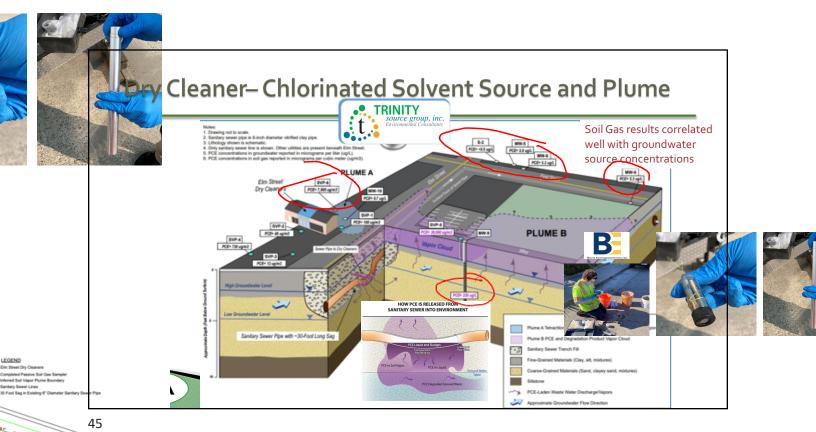
samplers to delineate two plumes related to chlorinated solvents across a residential neighborhood.

### Benefits: Low profile, public right

of way Access, no lawyers for Access Agreements.

CALIFORNIA





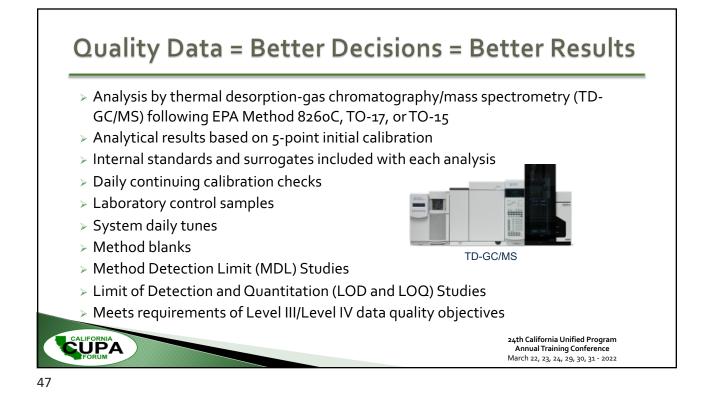
# Steady-State Sampling with Passive Soil Gas Samplers

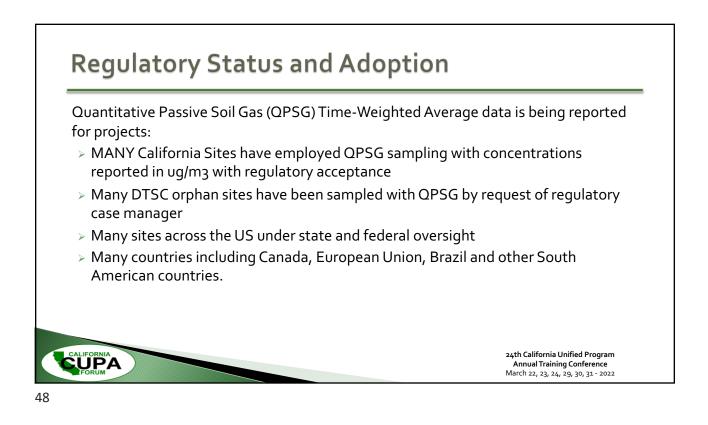
Passive soil gas samplers allow for the steady-state collection of samples over hours, days or weeks to measure organic compounds in indoor air, ambient air, sewer lines, and soil gas. Data are reported as average concentrations collected over time and are more representative of both short- and long-term health risks.

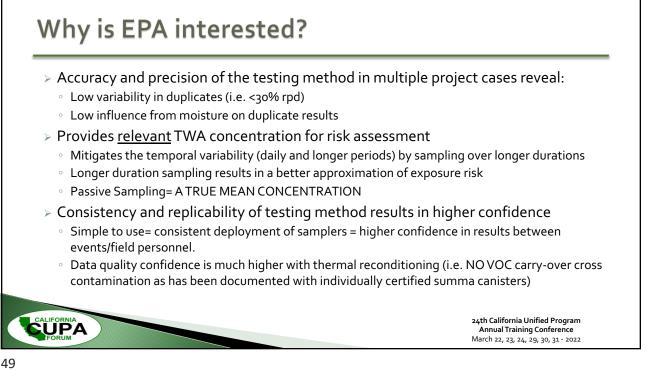
- Adsorbent media is hydrophobic. Allowing for accurate sampling in high moisture environments.
- Passive soil gas samplers provide more accurate steady-state soil gas TWA data for risk assessments and site investigations
- Passive samplers are easy to use (quick installation and precise deployment)
- Sample periods range from hours to weeks
- No pumps or flow regulators required (minimal equipment or field errors)
- No risk of leaks, bypassing or sample dilution (no helium leak test required)
- State-of-the-art analytical procedures produce high quality data and low LODs
- <u>No VOC carry-over issues</u>. Thermal reconditioning of each sampler is performed at 20C higher than testing procedures before each deployment.
   Batch and individual certification.

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